Pressure-driven filling of liquid metal in closed-end microchannels

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(Received 13 May 2018; published 7 September 2018)

We observe unsteady flow behavior of liquid metal during a pressure-driven injection process into a closed-ended polydimethylsiloxane microchannel. Constant pressure is applied at the inlet to allow eutectic gallium-indium (EGaIn) to completely fill the porous microchannels. In contrast to open channels [M. D. Dickey et al., Adv. Funct. Mater. 18, 1097 (2008)], the flow exhibits a complex unsteady behavior with sudden random length jumps and time stops. However, with appropriate formulation of a suitable mathematical model with the system using (i) the permeability of polydimethylsiloxane to air, (ii) previous descriptions of the nature of the EGaIn surface oxide layer, and (iii) a key probabilistic approach, we show that the average quantities defining the quantume-like flow can be accurately predicted. The proposed probabilistic formulation provides for the first time a description of the dynamics of the surface oxide layer, the breaking and healing characteristic times when EGaIn is driven in a microchannel. Importantly, this work provides a better understanding of complex flow behavior and lays the foundation for future work.

DOI: 10.1103/PhysRevE.98.032602

I. INTRODUCTION

Fluid displacement in porous media is of great importance in many diverse fields of science and engineering such as petroleum displacement in rocks [1], resin transfer molding of composite material production [2], and manipulation of liquid metals in microelectromechanical systems (MEMS) and microfluidic systems [3]. The manipulation of liquid metals has great potential for creating components in MEMS and microfluidic systems, such as switches [4], pumps [5], sensors [6], electrodes [7], and reflectors [8]. This is due to the fact that liquid metal is in the liquid state at room temperatures. Gallium alloys such as eutectic gallium-indium (EGaIn) are now the mostly used liquid metals in MEMS and microfluidic systems because of their low toxicity, low viscosity, and low vapor pressure compared to toxic liquid metals such as mercury. One of the important applications of EGaIn is the formation of electrodes in microfluidic systems [9,10], which is now being commonly used as an interdigital transducer to generate acoustic waves. However, despite decades of extensive research, the dynamics of the filling of liquid metal into a microchannel is still poorly understood.

For common liquids such as water, we have demonstrated in a separate work that their flow into a closed-end microchannel driven by a constant pressure [11] is a smooth and predictable process. Water enters the closed-end microchannel continuously and smoothly. This process can be modeled using air diffusion through a porous polydimethylsiloxane (PDMS) wall. In this article, we show that the filling of EGaIn driven by a constant pressure into a closed microchannel is very different from the process for water. The flow of EGaIn experiences total stops and sudden jumps: the flow stays halted for a while, and the liquid column then jumps to a new position and stops again. The words “jump” and “stop” aim to reflect the completely different time scales of the observed flow dynamics [12]. To understand the physics underlining this interesting “stop and jump” phenomenon, we propose here a mathematical model using gas diffusion and a random pressure component. The component is aimed to reflect the stochastic behavior observed. Once a very limited number of free physical parameters associated with that surface interaction was adjusted, we found a good agreement and consistency between the experimental observations and the model predictions. We hypothesize that this interesting phenomenon should be mainly attributed to the formation of a breakable oxide layer when EGaIn is exposed to air [13]. Although this oxide layer provides good mechanical stability to the developed metallic structures in microchannels [5,9,13], its formation and cracking while the liquid penetrates into the microchannel results in an unsteady flow [3,14]. While some innovative works have focused on complex fluid behavior on porous media [15–18], to the best of our knowledge, little or no theoretical analysis has been done before to characterize the dynamics and unsteady flow behavior of EGaIn in a closed-end porous microchannel, even though similar phenomena should be omnipresent in everyday technical applications such as the flow of molten metals in intricate molds (porous or not) and the flow of molten lava through cracks in the rocks. In this work we show the profound differences found from the unrestricted flow of a common liquid into a microchannel.

II. FLOW OF LIQUID METAL IN CLOSED MICROCHANNELS

Figure 1 shows a schematic of the experimental setup. The PDMS microchannels were fabricated using an SU8 mold (channel width \( w = 200 \mu m \), average height \( h \sim 47 \mu m \), and...
channel length $L = 2750 \mu$m) using standard photo and soft lithography technique [19]. A pressure controller (OB1 MK3; Elveflow) was connected to the inlet to provide a constant driving pressure [11,20]. EGaIn (No. 495425) was purchased from Sigma Aldrich and used directly. The PDMS device was mounted on an inverted microscope (Nikon Ti-E; Japan) and images were captured using a monochrome camera (Miro 3; Vision Research) at a rate of 10 frames per second. The low frame rate was used to capture the entire filling process. We kept the outlet of the channel closed to allow complete filling of EGaIn in all the microchannels. Complete filling is not possible with an open channel. EGaIn filled the microchannels under different constant driving pressures ($\Delta P = 1200$ and 1600 mb, respectively) imposed at the inlet.

We theoretically model the filling process of EGaIn in closed-end microchannels with the following assumptions: (i) air can diffuse throughout the channel wall due to the (molecular) porosity of PDMS; and (ii) the diffusion time of momentum ($t_D = hw/\nu$ is the kinetic viscosity coefficient of the liquid) is much shorter than the diffusion time of air into PDMS ($t_D = hw/D_{gas} = 4.7$ s, where $D_{gas}$ is the diffusion coefficient of air into PDMS, and $D_{gas} \approx 4.5 \times 10^{-9}$ m$^2$s$^{-1}$). $\Delta P$ is the imposed pressure by the controller, where the absolute (total) gas pressure is $\Delta P + P_{atm}$, and $P_{atm}$ is the atmospheric pressure. We also assumed that the liquid has essentially the same pressure everywhere and, thus, equals the total pressure imposed by the pressure controller. $\Delta P + P_{atm}$, We propose a dimensionless number, $G = 12\mu D_{gas} L^2/(w h^3 (P_{atm} + \Delta P))$, to measure the relative significance of the viscous relaxation time along the channel for a given pressure to the time of diffusion into PDMS. In our experiment, the dimensionless number is estimated as $G \sim 10^{-4}$. This inevitably implies that gas diffusion dominates over the viscous flow effect. We emphasize that given the sizes of our channels, the current analysis is very different from the force analysis which is often used for the case of capillary filling, where one may find experimental studies with $G \sim 10^6$ [21]. In brief, our model captures the unsteady random flow behavior of liquid metal in microchannels using subsequent steps of two metastable equilibrium states between (i) the escape of air from the porous PDMS while the metal interface is “frozen” by the formation of that oxide layer and (ii) the corresponding rapid spatial uptake of the liquid metal when that layer cracks due to the relative pressure buildup in the metal phase with respect to the remaining air (given previous air escape or degassing), leading to new metal surface exposed to gas (oxygen).

Figure 2 depicts the relative filling length $L_f/L$ of the random unsteady flow behavior of EGaIn under constant pressures of 1200 and 1600 mb, where $L_f$ is the filling length and $L$ is the total length of a microchannel. The data from nine adjacent microchannels were captured and analyzed using a customized Matlab image processing program. Figure 2 clearly shows that the filling dynamics reflects a complex and intricate mechanism. For example, the average filling speed, the number of “step” periods, and the widths of each step period are random and unpredictable. In this process, the trapped gas pressure in the unfilled part of the channel undergoes rapid increases due to the sudden advance of the liquid column followed by a relatively slow decrease due to degassing. This occurs by diffusion of gas through the PDMS, which dominates the penetration of liquid into the channels. The diffusion process is described and modeled in the Appendix.

Besides, the advancement of EGaIn in the microchannel of constant section $h \times w$ is governed by the equation

$$h w \frac{d \rho_a l}{dt} + l \int_{s_f} (0.79 \rho_{N_2} \nu_d N_2 + 0.21 \rho_{O_2} \nu_d O_2) ds = 0,$$

with the initial condition $l(0) = L$, where $\rho_a = 0.79 \rho_{N_2} + 0.21 \rho_{O_2}$ is the density of air trapped in the channel at pressure $P$, $l$ is the length of trapped air in the channel, and $s_f$ is the perimeter of the cross section of the channel. For a slender channel of constant cross section, one can assume an average gas diffusion $\bar{\nu}_d$ along the perimeter such that

$$\int_{s_f} (0.79 \rho_{N_2} \nu_d N_2 + 0.21 \rho_{O_2} \nu_d O_2) ds = 2(h + w) \rho_a \bar{\nu}_d,$$

taking as a good approximation $\rho_a \approx \rho_{N_2} \approx \rho_{O_2}$, where $\bar{\nu}_d = -D_{gas} \frac{\partial Y}{\partial r = D/2} = -2 \frac{D_{gas} \chi \Delta P}{\partial Y_{PDMS} = \partial Y_{O_2}} \frac{\partial Y_{O_2}}{\partial Y_{N_2}} + Y_0 \chi \psi \Delta P + Y_0$ the mass outflow of the gas by Fick’s first law. Here, we have defined the nondimensional variables $\eta = 2r/D$, $\psi = (P_{PDMS} - P_{atm})/\Delta P$, and $Y = \chi (P_{PDMS} - P_{atm}) + Y_0 \chi \psi \Delta P$ the mass fraction of air in PDMS, $\chi = 0.79 \rho_{N_2} + 0.21 \rho_{O_2}$ is the average permeability of PDMS to air. For simplicity, we define an equivalent cylindrical channel with diameter $D$ as $\pi D = 2(h + w)$ (see

FIG. 1. Experimental setup: (a) microchannels; (b) image of the filling process; (c) three-dimensional schematic of the PDMS channel.

FIG. 2. Experimental measurements of the nondimensional length of the EGaIn-filled part of each channel in the system as a function of time, under an imposed inlet pressure of (a) 1200 mb and (b) 1600 mb. Each color represents the filling process in one of the nine microchannels considered. Plots illustrate the complexity of the global evolution due to the strong stochastic component present in the evolution of each individual microchannel.
Appendix. Equation (1) can be rewritten as
\[
\frac{hw}{2\pi D \chi} \frac{1}{\rho l} \frac{d(\rho l)}{dt} = f(\tau) = 0,
\]
where \( \frac{d\psi}{d\tau} \big|_{\tau=1} = f(\tau) \) (Appendix). We define \( t_0 = D^2/(4D_{gas}) = (h + w)^2/(\pi^2 D_{gas}) \) as the characteristic diffusion time, and \( \tau = t/t_0 \). Solving Eq. (1) leads to
\[
\rho_l = \rho_a(0)l(0) \cdot \exp\left[ \alpha \int_0^\tau f(\tau')d\tau' \right],
\]
where \( \alpha = \frac{2hw + w^2}{\pi hw} \chi D \). This solution governs the general liquid flow in porous PDMS microchannels, either smooth or steplike.

To resolve the present problem of the unsteady flow of EGaIn in microchannels, some considerations should be made:

(i) In general, the thermodynamic evolution of the gas in the channel is assumed to be isothermal, because the thermal relaxation time, \( t_r \sim (hw)/\alpha_e = 7 \text{ ms} \), (where \( \alpha_e = 1.9 \times 10^{-5} \text{ m}^2/\text{s} \) is the thermal diffusivity of air) is much smaller than \( t_0 \) \( (t_0 = 1.37 \text{ s} \) in our experiments). This would be the case when the liquid metal column is static during the random steplike unsteady flow behavior of EGaIn in a microchannel.

(ii) However, the sudden compressions taking place in the advance steps of the liquid are fast enough to invalidate this assumption.

(iii) The actual length of the channel where gas is flowing into PDMS can be significantly different from \( l(t) \), because the liquid metal does not easily wet the surface of the PDMS wall: the presence of a gas layer between the metal and the PDMS prevents wetting and is very often observed along the channel. Indeed, the same solid thin oxide crust that forms at the liquid surface in contact with the gas side of the channel is also formed along the whole channel surface, which allows gas trapping between the two solids (PDMS and oxide crust). This is observed from the experimental images and depicted in Fig. 1(c). Thus, we have experimentally learned that the total length of the surface through which gas flows into the PDMS should be approximately equal to the total length \( L \) of the channel.

(iv) EGaIn forms a convex curvature when filling the microchannels. However, this curvature does not stay exactly the same throughout the process. In fact, as EGaIn fills the microchannel, we hypothesize that its surface becomes partially solidified by the heterogeneous buildup of a metal oxide layer at the virgin metal interface as it forms [13]. This layer may momentarily anchor to the PDMS wall by friction or partial snagging. As a result, a momentary “stopping” behavior is observed in this case.

Hence, the EGaIn flows in random “steps” as illustrated in Fig. 3, showing the cycles of the pressure amplitude and its stochastic nature. Each \( i \) step can be split into two parts with radically different time scales:

(1) In the first part, the initial gas pressure in the unfilled (gas) side of the channel is \( P_{i,1} \), and the length of the filled side \( L_f \) initially remains constant during a time interval \( \Delta t_i \). The characteristic time of this process is comparable to \( t_D \sim hw/D_{gas} \), and the gas side, with length \( l_i = L - L_f \), degasses through the PDMS wall. Consequently, the pressure decreases until it reaches a (stochastic) level \( P_{0,i+1} \), at which the EGaIn oxide layer breaks. This breakup finalizes the first part and exposes a virgin metal surface to the oxidizing gas.

(2) The second part is characterized by a very fast advance of the liquid column in the channel, with a characteristic time comparable to \( t_2 \sim \left( \frac{hw^2}{D} \right)^{1/2} \sim 0.13 \times 10^{-3} \text{ s} \ll t_0 \gg t_D \). In our experiments, the liquid density is \( \rho_i = 0.25 \times 10^3 \text{ kg/m}^3 \), \( L \sim 10^{-3} \text{ m} \), \( \Delta P \sim 10^4 \text{ Pa} \), and \( t_2/t_D \sim 10^{-4} \). The viscosity of EGaIn is \( \mu_i = 2 \text{ mPa} \cdot \text{s} \), which yields a characteristic viscous diffusion time in our channel \( t_{vl} = \rho_i hw/\mu_i = 0.03 \text{ s} \gg t_2 \). Besides, the thermal conductivity of PDMS is approximately \( K_s = 0.15 \text{ W/(mK)} \), its density \( \rho_s = 970 \text{ kg/m}^3 \), and its specific heat \( C_{p,s} = 1460 \text{ J/kg} \cdot \text{K} \), which yields a thermal diffusion coefficient \( \beta_s = K_s/(\rho_s C_{p,s}) \sim 10^{-7} \text{ m}^2/\text{s} \) and a characteristic time for thermal diffusion in the PDMS \( t_s = hw/\beta_s = 0.1 \text{ s} \gg t_2 \). Then we assume that the rapid advance of the compression step takes place adiabatically and with no diffusion of mass or heat through the PDMS.

The length of the unfilled side suddenly decreases from \( l_i \), the gas becomes compressed, the column rapidly decelerates, and the quickly oxidized liquid metal front again snags onto the PDMS surface. This finalizes the second part of step \( i \) and initiates the next step, \( i + 1 \).

To capture the above description, solution (3) should be properly adopted in a stepwise fashion. First, for the degassing part of each \( i \) step under a constant length \( l_i \), using Eq. (4), the ratio of initial to final pressures \( P_{0,i+1}/P_{i,1} \) can be expressed as
\[
\frac{P_{0,i+1}}{P_{i,1}} = \exp\left[ \alpha_i \int_{t_{i-1}}^{t_i} f(\tau) d\tau \right],
\]
where \( \alpha_i = \frac{2hw + w^2}{\pi hw} (0.79 \chi_{N_2} + 0.21 \chi_{O_2}) \Delta P \). Assuming an average value of \( \langle f(\tau) \rangle = B < 0 \) for our particular device,
independent of the operating conditions, one can express
\[
\frac{P_{0,i+1}}{P_{1,i}} = \exp(\alpha_i B \Delta \tau_i).
\] (6)
The constant \( B \) \((<-0)\) shall be a fitting parameter to be calculated from experiments.

Second, according to the rapid adiabatic process, the gas pressures \( P_{0,i+1} \) and \( P_{1,i+1} \), just before and after the sudden compression (advancement), respectively, are related to the lengths at each \( i \) step as
\[
\left( \frac{l_{i+1}}{l_i} \right)^{\gamma} = \frac{P_{0,i+1}}{P_{1,i+1}},
\] (7)
where \( \gamma = 1.4 \) is the adiabatic index for air. Here, the pressure difference between the gas and the liquid should be a function of its surface tension, able to withstand the curvature of the liquid metal front. That curvature is maximum (minimum) if the surface tension, able to withstand the curvature of the liquid metal front. That curvature is maximum (minimum)

Similarly, for the compression part of step \( i \):
\[
\frac{P_{0,i+1}}{P_{1,i}} = \exp(\alpha_i B \Delta \tau_i) \simeq \frac{P_{\text{atm}} + \Delta P - P_{L,i,d}}{P_{\text{atm}} + \Delta P + P_{L,i-1,c}},
\] (8)
where \( P_L \sim \sigma / h \) is the static Laplace pressure at the interface between the gas and EGaln, and \( \sigma = 0.624 \) N/m is the surface tension of the virgin liquid EGaln [9]. We have distinguished the random Laplace-related pressure at the end of the degassing \( i \) process \( P_{L,i,d} \) from the random one after sudden compression in the previous step \( P_{L,i-1,c} \). On average, the Laplace overpressures are much lower than the applied pressure, and therefore one may approximately write Eq. (6), after the appropriate linearization:
\[
\exp(\alpha_i B \Delta \tau_i) \simeq 1 - \frac{P_{L,i,d} + P_{L,i-1,c}}{P_{\text{atm}} + \Delta P} \equiv 1 - X_i.
\] (9)

Similarly, for the compression part of step \( i \) [Eq. (7)], one can write
\[
\frac{P_{0,i+1}}{P_{1,i+1}} = \left( \frac{l_{i+1}}{l_i} \right)^{\gamma} \simeq 1 + \frac{P_{L,i,c} + P_{L,i,d}}{P_{\text{atm}} + \Delta P} \equiv 1 + X'_i.
\] (10)

The combined statistical variables \( X_i \) and \( X'_i \) can be measured from experiments for different applied pressures, using the measured values of the successive lengths \( l_i \) taken from experiments (videos) and Eq. (7). A few fundamental considerations should be made here:

1. Even though \( P_{L,i,c} \) and \( P_{L,i,d} \) may have different distributions, the measured distributions of \( X_i \) and \( X'_i \) should approximately match because \( P_{L,i,c} \) and \( P_{L,i,d} \) are intermixed (rifled) in \( X_i \) and \( X'_i \). In our experiments, we match both mean values, \( \langle X_i \rangle \) and \( \langle X'_i \rangle \), to calculate the fitting constant \( B \).

2. The experimental measurements of \( \tau_i \) and \( l_i \) should be carefully filtered to eliminate sources of small deviations due to noise (vibrations, etc.).

3. Degassing steps leading to very large leaps in time may yield overpressures which are not small compared to \( P + \Delta P \). Given that these events are rare, one can suppress them from the statistical analysis.

4. In this analysis, we have neglected any possible dynamical coupling among the different channels. However, we are conscious that this coupling exists in reality at a certain level, which produces a further stochastic restriction in the mathematical behavior of the tails of the probability density functions; yet, these restrictions do not affect the averages of the involved variables or the validity of present analysis. This is discussed later.

Figure 4 shows the statistical distributions of both \( X_i \) and \( X'_i \). Using the fitting constant \( B = -9 \), both distributions can be relatively well approximated (considering the significant experimental uncertainties and the limited number of experiments available) by a log-normal distribution with mean \( \mu = -4 \) and standard deviation \( \xi = 2 \).

Similarly, the compression part of Step \( i \) should be

In summary, the following fitting parameters are used in our model:

1. The value of \( B \), which physically reflects the (nondimensional) tendency of the gas to escape through the PDMS, is uniquely determined by the device geometry and its material (PDMS).

2. The parameters of the statistical distribution that fits the experimental distributions of random Laplace pressures reflect the randomness of the process.

Figure 5 presents a selected simulation based on the results from our model. We observe that the model finely captures the compression and degassing steps and the whole physics of the liquid metal filling process. A perfect fitting between the experimental data and the model is virtually impossible due to the unpredictable randomness of the liquid metal flowing in each individual microchannel. However, our model provides accurate predictions using the ensemble average method. An important observation here is that the simulated filling represents the behavior of independent channels. However, our particular system comprises nine relatively close channels arranged in parallel and all connected to a common feeding large channel (see Fig. 1): this provokes a measurable interaction among the different channels reflected in the phenomenon of synchronism, which will be studied in further works. In particular, we can suggest that if there were no synchronism, dividing the whole filling time interval into small sections (for example, of 0.1 s each), the coincidence of leaps within
a certain time section should follow a Poisson distribution. However, this is not the case for our system: the probability of finding simultaneous leaps in the same time section becomes orders of magnitude higher than that predicted by a Poisson distribution. Although this would not affect the average and standard deviations of the statistical distributions shown in Fig. 4, their shapes (i.e., higher-order moments of the distributions) could be significantly affected, as observed. Indeed, one may observe a nonnatural shape of the distribution of $X'_i$ around its median. This is why the experimental evolution shown in Fig. 2 exhibits much less dispersion than a simulation with a large number of runs; however, the average of those runs always yields the same asymptotic result, naturally.

The arithmetic average of the experimental data can be compared to the average results from the model. The fitting between both experimental results and the model with a single fitting constant and a fitting experimental statistical distribution is shown in Fig. 6. The simulated results come from the arithmetic average of the model results of 100 runs. Even though we use experimental data for only nine fingers, there is very good consistency between the simulated results and the experimental ones, for imposed inlet pressures of both 1200 and 1600 mb. Thus, our model and the corresponding fitting parameters capture the physics and explain satisfactorily the unsteady random flow of EGaIn in a microchannel of a porous matrix like PDMS.

In conclusion, the random unsteady flow behavior of EGaIn in porous microchannels under a constant driving pressure is theoretically investigated and experimentally validated. Complete filling of the porous microchannels can be realized by imposing a constant pressure at the inlet. We observed a random “stopping” and “jumping” phenomenon, which is attributed to the formation of a transient oxide layer that subsequently cracks and exhibits a new liquid surface, allowing the liquid column to penetrate into the porous microchannel. This new surface is oxidized again and eventually snags on the PDMS surface after a transient time, stopping the advance of the column. A theoretical model based on the gas diffusion and random surface tension overpressures is developed to describe this phenomenon. The physical properties and experimental data are well captured and described by our model. Our work offers the basic understanding of the unsteady random flow of liquid metals in porous microchannels prefilled with an oxidizing gas such as air.

ACKNOWLEDGMENTS

This work was partially performed at the Queensland Node of the Australian National Fabrication Facility, a company established under the National Collaborative Research Infrastructure Strategy to provide nano- and microfabrication facilities for Australia’s researchers. A.M.G.-C. acknowledges support from the Ministerio de Economía y Competitividad through Project No. DPI2016-78887-C3-1R. S.H.T. and N.-T.N. gratefully acknowledge the support of the Australian Research Council (ARC) Discovery Early Career Researcher Award (DECRA; No. DE170100600), Griffith University–Simon Fraser University and Griffith University–Peking University collaboration grant. W.G. and H.-D.X. gratefully acknowledge the support of the National Natural Science Foundation of China (through Grants No. 11472094, No. 11772259, and No. U1613227) and the 111 project of China (No. B17037).

APPENDIX: GAS DIFFUSION MODEL

IN THE PDMS MATRIX

Here, we propose an equivalent cylindrical channel, with diameter $D$, to approximate the rectangular microchannels. This equivalence is based on the same perimeters of two different geometries, which gives $2(\pi r < \pi D)$ or $D = 2(\pi r)$. In this way, the conversion provides simplicity when we model the system in polar coordinates. We assume that the air in the PDMS channel consists of two main components: nitrogen and oxygen. Then we investigate the filling process in the channel domain $(0 < r < D/2)$ and the gas diffusion in the PDMS domain $(r > D/2)$. In the following, $P_{\text{PDMS}}(r), Y(r)$, and $C(r)$ denote the pressure of air in the PDMS domain, the mass fraction of gas in the PDMS domain, and the volume fraction of gas in PDMS, respectively. Sadrzadhel et al. [22] reported that nitrogen and oxygen have similar permeabilities.
in PDMS, which can be expressed as
\[
\frac{\partial C_{N_i}}{\partial P} = 9.35 \times 10^{-7} \text{ Pa}^{-1}, \quad \frac{\partial C_{O_2}}{\partial P} = 1.87 \times 10^{-6} \text{ Pa}^{-1}.
\]

Equation (A1) leads to the permeabilities written in terms of the mass fractions of each gas component of air (the form used in this work for the gas permeabilities):
\[
\chi_{N_i} = \frac{\partial Y_{N_i}}{\partial \eta} = \rho_{N_i} \frac{\partial C_{N_i}}{\partial P} = 1.1 \times 10^{-9} \text{ Pa}^{-1},
\]
\[
\chi_{O_2} = \frac{\partial Y_{O_2}}{\partial \eta} = \rho_{O_2} \frac{\partial C_{O_2}}{\partial P} = 2.5 \times 10^{-9} \text{ Pa}^{-1}.
\]

For a given pressure \( P \) in the unfilled part of the channel (gas side), the mass fraction of the gas, either nitrogen or oxygen, in the PDMS domain can be calculated from (A2) as
\[
Y - Y_0 = \chi (P_{PDMS} - P_{atm}),
\]
where \( P_{PDMS} \) is the gas pressure in the PDMS, and \( \chi = 0.79 \chi_{N_i} + 0.21 \chi_{O_2} \) is the average permeability of PDMS to air. According to Fick’s second law, the diffusion of gas into PDMS is governed by
\[
\frac{\partial Y}{\partial t} = D_{\text{gas}} \frac{1}{r} \frac{\partial}{\partial r} \left( r \frac{\partial Y}{\partial r} \right).
\]

In a smooth filling process, such as that of water or ethanol, the air trapped is pushed by the liquid column at a constant pressure along the process (i.e., there is no “stopping” or “jumping”). Then the boundary conditions (BCs) in this case are

\[
Y(r = D/2, t) = Y(P_{PDMS} = P),
\]
\[
Y(r \gg D/2, t) = Y(P_{PDMS} = P_{atm}) = Y_0,
\]
\[
\frac{\partial Y}{\partial r}(r = D/2, t) = \chi \frac{dP}{dt}.
\]

Here \( D_{\text{gas}} \) is the diffusion coefficient of the gas in PDMS, which is almost the same for both \( N_2 \) and \( O_2 \): \( D_{\text{gas}} \approx 4.5 \times 10^{-9} \text{ m}^2 \text{s}^{-1} \) [22]. We define the nondimensional variables \( \eta = 2r/D, \quad \tau = t/t_0, \quad \psi = (P_{PDMS} - P_{atm})/\Delta P \), and \( Y = \chi (P_{PDMS} - P_{atm}) + Y_0 = \chi \psi \Delta P + Y_0 \), where \( t_0 = D^2/(4D_{\text{gas}}) = (h + w)^2/(\pi^2D_{\text{gas}}) \) denotes the characteristic diffusion time. Equation (A4) can be nondimensionalized as
\[
\frac{\partial \psi}{\partial \tau} = \frac{1}{\eta} \frac{\partial}{\partial \eta} \left( \eta \frac{\partial \psi}{\partial \eta} \right).
\]

The linear nature of Eq. (A5) allows its analytical solution with BCs to be decomposed as the sum of particular solutions,
\[
\psi(\eta, \tau) = \sum_i \psi_i(\eta, \tau),
\]
where \( \psi_i(\eta, \tau) \) denotes different solutions satisfying Eq. (A5) with generic BCs. Here, we consider three of those particular solutions. First, we introduce the simplest one,
\[
\psi_0(\eta, \tau) = A_0 \left( \frac{\eta^2}{4} + B \ln \eta + A_1 \right),
\]
which fulfills certain, but not the whole set of, BCs. To gain refinement with additional solutions, we can introduce a self-similar variable, \( \xi = \eta^2/(4\tau) \), which leads to
\[
\frac{\partial \psi}{\partial \tau} = \frac{d}{d\xi} \frac{\partial \psi}{\partial \xi} = -\eta \frac{d}{d\xi}, \quad \frac{\partial \psi}{\partial \eta} = \frac{d}{d\xi} \frac{\partial \psi}{\partial \eta} = \frac{\eta}{2\tau} \frac{d}{d\xi},
\]
and transforms Eq. (A5) into an ordinary equation,
\[
(\xi \psi')' + \xi \psi' = 0,
\]
with BCs
\[
\psi(\eta = 1, \tau) = 1, \quad \psi(\eta \gg 1, \tau) = 0, \quad \frac{\partial \psi}{\partial \tau}(\eta = 1, \tau) = \frac{t_0}{\Delta P} \frac{d(P - P_{atm})}{dt}.
\]

A generic analytical solution to Eq. (A8) is \( \psi = C \exp(-\xi) \), where \( \xi = \psi' \). Thus we have
\[
\psi_i(\xi) = C \exp(-\xi),
\]
where \( Ei(x) \) is the exponential integral function, expressed as \( Ei(x) = -\int_{-\infty}^{x} \frac{e^{-t}}{t} \, dt \). Again, this solution does not completely resolve the whole set of BCs. Finally, using the method of separation of variables, Eq. (A5) can also be solved as
\[
\psi_2(\eta, \tau) = \int_{\infty}^{\tau} A(\lambda) \exp(-\lambda \tau) \cdot K_0(\lambda \eta) d\lambda,
\]
where \( K_0 \) is the modified Bessel function of order 0, and \( A(\lambda) \) reflects the solution to the set of BCs not fulfilled by \( \psi_0 \) or \( \psi_1 \). In summary, the analytical solution to Eq. (A5) can be, finally, written as the sum of (A7), (A9), and (A10),
\[
\psi(\eta, \tau) = \psi_0(\eta, \tau) + \psi_1(\xi) + \psi_2(\eta, \tau) + \ldots,
\]
where \( A_0, A_1, A(\lambda), B, \) and other coefficients are solved to meet the initial and boundary conditions of gas diffusion in PDMS. In a smooth filling process, the pressure in the unfilled (gas) part of the channel is constant, which means that \( g(\tau) = 0 \) and results in
\[
\frac{d(\tau)}{f(\tau)} = B + \frac{C}{\tau} \cdot (1 + 2\tau) \cdot \exp \left( -\frac{1}{4\tau} \right) + \int_{\infty}^{\tau} A(\lambda) \exp(-\lambda \tau) [K_0(\lambda) - K_1(\lambda)] d\lambda,
\]
where \( K_0 \) is the modified Bessel function of order 1. However, in the case of a steplike evolution (liquid EGaIn) the gas pressure is not constant in each degassing step at a constant length. The BCs in this case are
\[
\left. \frac{\partial Y}{\partial r} \right|_{r=D/2} = \frac{\pi \chi \Delta P}{h + w} \left. \frac{\partial \psi}{\partial \eta} \right|_{\eta=1} = \frac{\pi \chi \Delta P}{h + w} f(\tau),
\]
\[
\left. \frac{\partial Y}{\partial \tau} \right|_{r=D/2} = \frac{\chi \Delta P}{t_0} \left. \frac{\partial \psi}{\partial \tau} \right|_{\eta=1} = \frac{\chi \Delta P}{t_0} g(\tau),
\]
where \( f(\tau) = \frac{d\psi}{d\eta} \bigg|_{\eta=1} \) and \( g(\tau) = \frac{d\psi}{d\tau} \bigg|_{\eta=1} \), respectively. A general solution of this problem cannot be analytically obtained and should be numerically resolved for each particular
configuration. However, some simplifying hypotheses can be made on $f(\tau)$ such that it can be approximately obtained from experiments for each given configuration.

In summary, $f(\tau)$ is the key factor which determines the degassing process through PDMS that leads to the advancement of the liquid column.